

Basic Chemical Segregation

CLASS OF CHEMICALS	RECOMMENDED STORAGE METHOD	EXAMPLES	INCOMPATIBLES *
Carcinogens	Label all containers as "Cancer Suspect Agents." Store according to the hazardous nature of the chemical, using the appropriate security when necessary.	Benzidine, beta-naphthylamine, benzene, methylene chloride, beta-propiolactone	See the original chemical label or the SDS for more information.
Compressed Gases - Flammable	Store in a cool, dry area, away from oxidizing gases. Securely strap or chain cylinders to a wall or bench top.	Methane, acetylene, propane	Oxidizing and toxic compressed gases, oxidizing solids
Compressed Gases - Oxidizing	Store in a cool, dry area, away from oxidizing gases. Securely strap or chain cylinders to a wall or bench top.	Oxygen, chlorine, bromine	Flammable gases
Compressed Gases - Poisonous	Store in a cool, dry area, away from oxidizing gases. Securely strap or chain cylinders to a wall or bench top.	Carbon monoxide, hydrogen sulfide (H ₂ S)	Flammable and/or oxidizing gases
Corrosives - Acids	Store in a separate acid storage cabinet. Segregate oxidizing acids (i.e., Chromic, nitric, sulfuric, perchloric acids) from organic acids	Mineral acids – hydrochloric acid, sulfuric acid, nitric acid, perchloric acid, chromic acid, chromerge	Flammable liquids, flammable solids, bases, oxidizers
Corrosives - Bases	Store in a separate storage cabinet.	Ammonium hydroxide, sodium hydroxide	Flammable liquids, oxidizers, poisons, acids
Flammable Liquids	Store in a grounded flammable storage cabinet.	Acetone, benzene, diethyl ether, methanol, ethanol, toluene, glacial acetic acid	Acids, bases, oxidizers, poisons

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Flammable Solids	Store in a separate dry, cool area away from oxidizers, corrosives, and flammable liquids.	Phosphorus (yellow), calcium carbide, picric acid, benzoyl peroxide	Acids, bases, oxidizers, poisons
General Chemicals Non-Reactive	Store on general laboratory benches or shelving preferable behind glass doors, or below eye level.	Agar, sodium chloride, sodium bicarbonate, most non-reactive salts	See the original chemical label or the SDS for more information.
Oxidizers	Store in a spill tray inside a non-combustible cabinet, separate from flammable and combustible materials.	Sodium hypochlorite, benzoyl peroxide, potassium permanganate, potassium chlorate, potassium dichromate The following are generally considered oxidizing substances: peroxides, perchlorates, chlorates, nitrates, bromates, superoxides.	Separate from reducing agents, flammables, combustibles
Peroxide-Forming Chemicals	Store in air-tight containers in a dark, cool, dry area. (See Suggested Storage Time Limits for Common Peroxidizable Compounds for more information.)	Diethyl ether, acetaldehyde, acrylonitrile	See the original chemical label or the SDS for more information.
Poisons/Toxic Compounds	Store separately in a vented, cool, dry area in unbreakable chemically resistant secondary containers.	Cyanides, heavy metal compounds (i.e., cadmium, mercury, osmium)	Flammable liquids, acids, bases, oxidizers
Shock Sensitive Materials	Store in a secure location away from all other chemicals.	Ammonium nitrate, nitro urea, picric acid (in a dry state), trinitroaniline, trinitroanisole, trinitrobenzene, trinitrophenol/picric acid, trinitrotoluene, urea nitrate, zirconium picramate	Flammable liquids, oxidizers, poisons, acids, bases

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Strong Reducing Agents	<p>Store in a cool, dry, well-ventilated location.</p> <p>Most of these are water-reactive.</p> <p>Segregate from all other chemicals.</p>	Acetyl chloride, thionyl chloride, maleic anhydride, ferrous sulfide	See the original chemical label or the SDS for more information.
Teratogens	<p>Label all containers as "Suspect Reproductive Hazard."</p> <p>Store according to the hazardous nature of the chemical, using the appropriate security when necessary.</p>	Lead, mercury compounds, benzene, aniline	See the original chemical label or the SDS for more information.
Water Reactive Chemicals	<p>Store in a dry, cool location.</p> <p>Protect these chemicals from the water fire sprinklers.</p>	Sodium metal, potassium metal, lithium metal, lithium aluminum hydride	Separate from all aqueous solutions and oxidizers

This information is from the **Environment, Health, and Safety Online** website (EHSO). We wish to give them full credit and our thanks for this information.

***Read the chemical label or the SDS if you have questions or call OH&S.**

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